

NILASAILA INSTITUTE OF SCIENCE & TECHNOLOGY SERGARH-756060, BALASORE (ODISHA) (Approved by AICTE& affiliated to SCTE&VT, Odisha)



LESSON PLAN

SUBJECT: Th.2b. ENGINEERING CHEMISTRY

CHAPTER WISE DISTRIBUTION OF PERIODS

		No. of	
Sl.No.	Name of the chapter as per the Syllabus	Periods	
		as per	periods
		the	actually
		Syllabu	needed
		S	
1	Physical Chemistry	22	22
1	i nysicai chemistry		
2	Inorganic Chemistry	8	8
3	Organic Chemistry	10	10
	organic enemistry	10	10
4	Industrial Chemistry	20	20
	,		
	Total Period:	60	60
4	Industrial Chemistry Total Period:		

Discipline: ALL BRANCHES	YEAR: 1 ST	Name of the Teaching Faculty: RAJAT KUMAR SAHOO	
Week	Class Day	Theory / Practical Topics	
1 st	1 st	Chapter 1: Atomic structure : Fundamental particles (electron, proton & neutron Definition, mass and charge).	
	2 nd	Atomic mass and mass number, Definition, examples and properties of Isotopes, isobars and isotones.	
	3 rd	Rutherford's Atomic model (postulates and failure),	
	4 th	Bohr's Atomicmodel (Postulates only), Bohr-Bury scheme,	
2 nd	1 st	Aufbau's principle, Hund's rule	
	2 nd	Electronic configuration (up to atomic no 30).	
	3 rd	QUESTION AND ANSWER DISCUSSION	
	4 th	Chapter 2 : Chemical Bonding : Definition , types (Electrovalent, Covalent and Coordinate bond with examples	
3 rd	1 st	(formation of NaCl, MgCl2, H2,Cl2, O2, N2, H2O, CH4, NH3, NH4 +, SO2).	
	2 nd	QUESTION AND ANSWER DISCUSSION	
	3 rd	Chapter 3 : Acid base theory : Concept of Arrhenius, Lowry Bronsted	
	4 th	Lewis theory for acid and base with examples (Postulates and limitations only). Neutralization of acid & base.	
4 th	1 st	Definition of Salt, Types of salts (Normal, acidic, basic, double, complex and mixed salts, definitions with 2 examples from each).	
	2 nd	QUESTION AND ANSWER DISCUSSION	
	3 rd	Chapter 4: Solutions : Definitions of atomic weight, molecular weight, Equivalent weight.	
	4 th	Determination of equivalent weight of Acid, Base and Salt.	
5 th	1 st	Modes of expression of the concentrations (Molarity , Normality & Molality) with Simple Problems.	
	2 nd	pH of solution (definition with simple numericals) Importance of pH in industry (sugar, textile, paper industries only)	
	3 rd	QUESTION AND ANSWER DISCUSSION	
	4 th	Chapter 5 : Electrochemistry : Definition and types (Strong & weak) of Electrolytes with	
	1 st	Electrolysis (Principle & process) with example of NaCl (fused and aqueous solution).	

+h	2 nd	Faraday's 1st and 2nd law of Electrolysis (Statement, mathematical expression and Simple numerical)		
6 th	3 rd	Industrial application of Electrolysis- Electroplating (Zinc only).		
	4 th	QUESTION AND ANSWER DISCUSSION		
	1 st	Chapter 6: Corrosion: Definition of Corrosion, Types of Corrosion- Atmospheric Corrosion, Waterline corrosion. Mechanism of rusting of Iron only.		
7 th	2 nd	Protection from Corrosion by (i) Alloying and (ii) Galvanization.		
	3 rd	QUESTION AND ANSWER DISCUSSION		
	4 th	Chapter 7 : Metallurgy: Definition of Mineral, ores , gangue with example. Distinction between Ores And Minerals		
8 th	1 st	General methods of extraction of metals, i) Ore Dressing ii) Concentration (Gravity separation, magnetic separation, Froth floatation & leaching)		
	2 nd	iii) Oxidation (Calcinations, Roasting) iv) Reduction (Smelting, Definition & examples of flux, slag) v) Refining of the metal (Electro refining, & Distillation		
	3 rd	QUESTION AND ANSWER DISCUSSION		
	4 th	EXAM		
9 th	1 st	Chapter 8 : Alloys: Definition of alloy. Types of alloys(Ferro, Non Ferro & Amalgam) with example.		
	2 nd	Composition and uses of Brass, Bronze, Alnico, Duralumin		
	3 rd	QUESTION AND ANSWER DISCUSSION		
	4 th	Chapter 9: Hydrocarbons: Saturated and Unsaturated Hydrocarbons (Definition with		
	1 st	Aliphatic and Aromatic Hydrocarbons (Huckle's rule only). Difference between Aliphatic and aromatic hydrocarbons.		
10 th	2 nd	IUPAC system of nomenclature		
	3 rd	Alkane, Alkene, Alkyne,		
	4 th	alkyl halide and alcohol .		
	1 st	Uses of some common aromatic compounds (Benzene, Toluene, BHC, Phenol, Naphthalene, Anthracene and Benzoic acid) in daily life.		
11 th	2 nd	QUESTION AND ANSWER DISCUSSION		
	3 rd	Chapter 10: Water Treatment: Sources of water, Soft water, Hard water, hardness, types of Hardness (temporary or carbonate and permanent or non-		
	4 th	Removal of hardness by lime soda method(hot lime & cold lime—Principle, process & advantages) , Advantages of Hot lime over cold lime process.		

	1 st	Organic Ion exchange method (principle, process, and regeneration of exhausted		
12 th	-	resins)		
	2 nd	QUESTION AND ANSWER DISCUSSION		
	3 rd	Chapter 11 : Lubricants: Definition of lubricant, Types (solid, liquid and semisolid with		
	4 th	specific uses of lubricants (Graphite, Oils, Grease), Purpose of lubrication		
13 th	1 st	QUESTION AND ANSWER DISCUSSION		
	2 nd	Chapter 12 : Fuel: Definition and classification of fuel, Definition of calorific value of fuel, Choice of good fuel		
	3 rd	Liquid: Diesel, Petrol, and Kerosene Composition and uses.		
	4 th	Gaseous: Producer gas and Water gas (Composition and uses). Elementary idea about LPG, CNG and coal gas (Composition and uses only).		
14 th	1 st	QUESTION AND ANSWER DISCUSSION		
	2 nd	Chapter 13: Polymer: Definition of Monomer, Polymer, Homo-polymer, Co-polymer and Degree of polymerization.		
	3 rd	Difference between Thermosetting and Thermoplastic, Composition and uses of Polythene, & Poly-Vinyl Chloride and Bakelite		
	4 th	Definition of Elastomer (Rubber). Natural Rubber (it's draw backs).		
15 th	1 st	Vulcanisation of Rubber.Advantages of Vulcanised rubber over raw rubber.		
	2 nd	QUESTION AND ANSWER DISCUSSION		
	3 rd	Chapter 14: Chemicals in Agriculture: Pesticides: Insecticides, herbicides, fungicides- Examples and uses.		
	4 th	Bio Fertilizers: Definition, examples and uses.		